Assignment for the Infrared Spectrum of Solid Sodium Propionate from Low-Temperature Measurements in Combination with ¹³C Isotopic Shifts

Masato Kakihana and Tadashi Nagumo Department of Chemistry, The National Defense Academy, Yokosuka, Japan

Z. Naturforsch. 42 a, 477 – 484 (1987); received January 14, 1987

The infrared spectra of CH₃CH₂COONa and its ¹³C-labeled modifications (1-¹³C, 2-¹³C, and 3-¹³C) suspended in KBr disks were measured in the region 4000 – 200 cm⁻¹ at room temperature and liquid nitrogen temperature. Overlapping complex band contours appeared in some regions of the room temperature spectrum, most notably in the region 1500 – 1350 cm⁻¹, where 5 fundamentals having contributions from the methyl deformation, methylene bending, and carboxylate stretching modes should occur. In contrast to this, excellent resolution was reached at the low temperature, from which all 22 fundamentals expected in the whole spectral region investigated were detected. A complete assignment of the fundamentals is proposed mainly on the basis of the characteristic isotopic shifts of the three ¹³C substituted sodium propionate species. A fair number of the fundamentals were found to feature coupled modes having contributions from several group vibrations.

Introduction

Propionate ion is important not only in organic chemistry but also in inorganic chemistry as one of the simplest ligands forming metal complexes. To our knowledge, however, no assignment for the infrared spectrum of the ionized form of propionic acid [1] has been reported so far. In recent papers [2-4]relating to the vibrational analyses of sodium acetate and pyruvate, we have shown that i) lowtemperature infrared spectra make it possible to clarify regions of the spectra which are overlapped at room temperature, and thus to locate the fundamental frequencies without ambiguity, and that ii) the frequencies from heavy atom isotope (13C, 18O, etc.) labeled modifications rather than D-labeled species allow for more reliable assignments of the fundamentals.

In the present work, the infrared spectra of CH₃CH₂COONa (parent species), CH₃CH₂¹³COONa (1-¹³C), CH₃¹³CH₂COONa (2-¹³C), and ¹³CH₃CH₂-COONa (3-¹³C) were measured at 80 and 290 K. Detailed profiles of the low-temperature spectra are given in order to demonstrate their great advantage in reaching excellent resolution. Vibrational assignments are presented for all the fundamentals observed in the spectral region 4000 – 200 cm⁻¹ with the aid of the ¹³C shifts of the fundamentals.

Reprint requests to Dr. M. Kakihana, Department of Chemistry, The National Defense Academy, Hashirimizu 1-10-20, Yokosuka 239, Japan.

Experimental

Potassium bromide and CH₃CH₂COONa were obtained commercially (E. Merck Co. Ltd., Darmstadt). The ¹³C-enriched modifications of sodium propionate were purchased from Merck Sharp and Dohme Canada Ltd. with 99% ¹³C isotopic purity for 1-13C and with 90% for 2 and 3-13C. The handling of the samples and the preparation of KBr disks were as described in [2]. The infrared spectra were measured in the frequency regions of $4000 - 450 \text{ cm}^{-1}$ and $450 - 200 \text{ cm}^{-1}$ by using a Bruker IFS-113v FT-IR spectrometer with a resolution of 0.5 cm⁻¹ and a Jasco Model A-702 spectrometer with slit program N (spectral resolution 2.5 cm⁻¹), respectively. Liquid nitrogen cryostats with KRS-5 windows, Model CF-1104 (Oxford Instrument) and Model TM-1-511 (Cryogenic Research Laboratory), were used for the low-temperature measurements. The band frequencies lying in the region $450 - 200 \text{ cm}^{-1}$ were calibrated against the standard absorption bands of gaseous H₂O [5].

Results and Discussion

The great advantage of measuring spectra at low temperature is demonstrated for the parent species in Figs. 1 and 2 covering the spectral regions $3040-2880~\rm cm^{-1}$ and $1700-250~\rm cm^{-1}$, respectively. The ν_2 band in Fig. 1 and the ν_5 , ν_6 , and ν_{18} bands in

0932-0784 / 87 / 0500-0477 \$ 01.30/0. - Please order a reprint rather than making your own copy.



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung "Keine Bearbeitung") beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen. On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition "no derivative works"). This is to allow reuse in the area of future scientific usage.

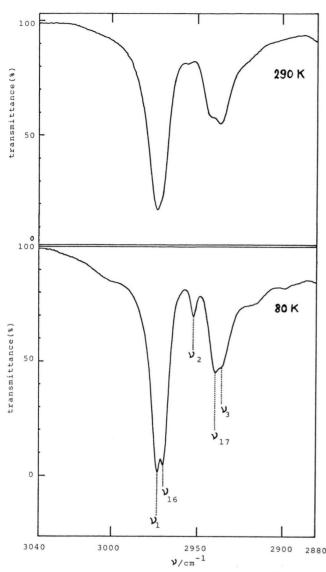


Fig. 1. Characteristic absorption bands due to CH₃- and CH₂-stretching modes of CH₃CH₂COONa at 80 (below) and 290 (above) K; 0.35 mg of sample suspended in 150 mg of KBr. The symbols v_i (i = 1-3, 16, 17) correspond to those given in Table 1.

Fig. 2, which were hardly discernible in the room temperature spectra, clearly showed up in the low temperature spectra. Such a sharpening effect is also seen for the v_{19} , v_{22} and v_{15} bands in Figure 2. Some of the lines at 80 K consist of two or more individual absorption lines. These lines presumably result from different crystal sites of sodium propionate. This interpretation is supported by an often similar structure of the corresponding absorption lines in

¹³C-labeled modifications of sodium propionate. Low temperature spectra are used throughout in this paper for the identification of the absorption lines.

In the present vibrational assignments, all the isotopic species concerned are assumed to belong a symmetry point group C_s in which the vibrational representation has the form 15 A' + 9 A''. Since the methyl and CO2 torsional A" modes are expected to be observed in a low frequency region (less than 200 cm⁻¹), they were not investigated in this paper. The fundamental frequencies observed at 80 K, along with their proposed assignments, are listed in Table 1. The initial assignments of the fundamentals were carried out on the basis of the qualitative empirical rules available [6] and of comparisons with the vibrational assignments on related molecules such as sodium acetate [3], pyruvate [4], and propane [7]. The present ¹³C isotopic shifts of the fundamentals are quite useful to assess the proposed assignments as indicated below.

(1) CH-stretching region

In the region of 3000 cm⁻¹, a total of 5 fundamentals are expected to be observed; three of them are classified as CH3 stretching modes, the others as CH₂ stretching vibrations. An expanded section of the low temperature spectrum for the parent species is shown in Fig. 3, together with the corresponding spectra for the ¹³C-labeled modifications. Going from the high frequency side to the lower frequency side, one finds a strong intensity band with a doublet structure (v_1 and v_{16}), a weak intensity band (v_2) , and a medium intensity band (v_{17}) with a shoulder (v_3) at lower frequency (see Figure 3 a). The doublet could be assigned to the near degenerate antisymmetric CH3 stretching modes, since the ¹³C-substitution at the methyl carbon caused a significant lower frequency shift (ca. 8 cm⁻¹) of the band in question (see Figure 3 d). The v_2 and v_{17} bands observed at 2952.5 and 2940.0 cm⁻¹ for the parent species could be due to the symmetric and antisymmetric stretching motions of the CH₂ group. The corresponding fundamentals for the 2-13C labeled modification were found at 2943.5 and 2933.3 cm⁻¹ with lower frequency shifts (see Figure 3c). The remaining symmetric CH₃ stretching fundamental for the parent species is submerged under the v₁₇ band assigned to the CH₂ antisymmetric stretching mode, and it was observed only as

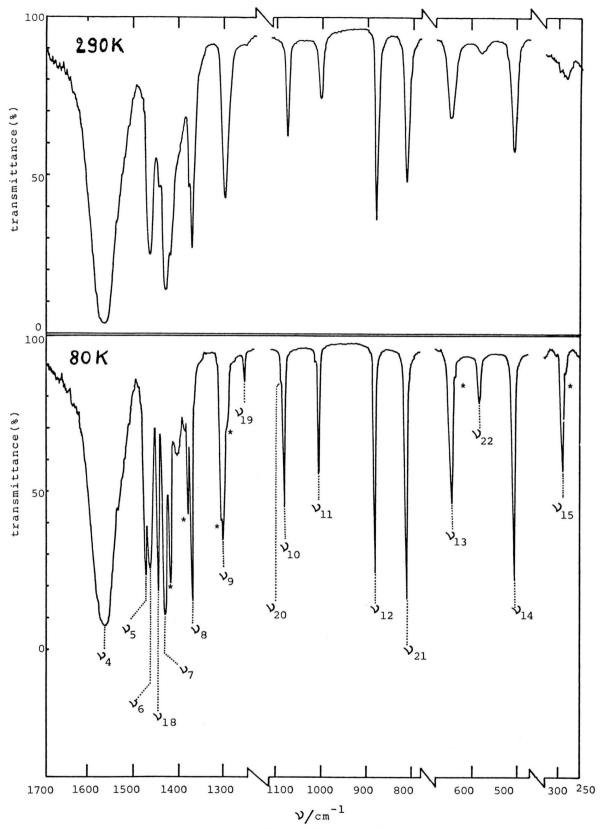


Fig. 2. Selected regions of the infrared spectrum of CH₃CH₂COONa at 80 (below) and 290 (above) K; 0.35 mg of sample suspended in 150 mg of KBr. For the symbols v_i (i = 4-22) see the notations in Table 1. Bands in the spectrum at 80 K marked by an asterisk represent factor group splittings.

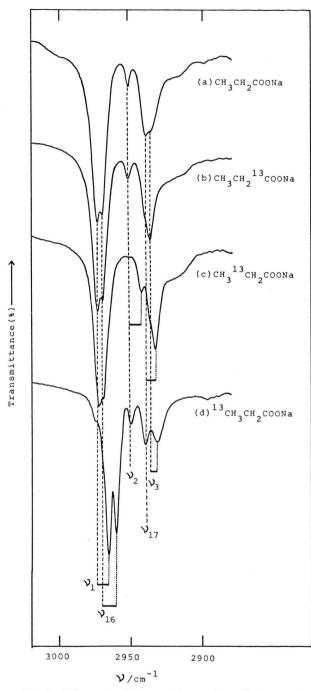


Fig. 3. CH₃- and CH₂-stretching region of the spectra of CH₃CH₂COONa (a) CH₃CH₂¹³COONa (b), CH₃¹³CH₂COONa (c), and ¹³CH₃CH₂COONa (d) at 80 K; 0.35 mg of sample suspended in 150 mg of KBr. Broken and dotted lines indicate positions of fundamentals for the parent and ¹³C-labeled species, respectively. Horizontal lines connecting broken and dotted lines represent magnitudes of ¹³C-shifts of fundamentals. For symbols v_i , see the notations given in Table 1.

a shoulder (v_3) at about 2937 cm⁻¹. The ¹³C-substitution at the methyl carbon caused a lower frequency shift (ca. 8 cm⁻¹) of the v_3 band, so that the fundamental due to the CH₃ symmetric stretching mode for the 3-¹³C labeled modification was separately observed from the v_{17} band as shown in Figure 3 d.

(2) CO₂-stretching and (HCH)-bending region

The band observed at $1563.2 \text{ cm}^{-1}(v_4)$ for the parent species is very intense as shown in Figure 2. The v_4 band shifts by ca. 40 cm^{-1} upon $^{13}\text{C-sub-}$ stitution of the carboxylate group, whereas the same band has a high degree of the antisymmetric CO2 of the methylene and methyl groups (see v4 in Table 1). The intensity of the band and the result of the 13 C isotopic shift strongly suggest that the v_4 band has a high degree of the antisymmetric CO₂ stretching character. In the low temperature spectrum for the parent species all 7 fundamentals (5A' + 2A'') expected in the spectral region 1500 to 1200 cm⁻¹ may be localized as shown in Figure 4. Of these, the assignment of the fundamental due to a symmetric CO₂ stretching motion is straightforward; the v_7 band at 1428.6 cm⁻¹ for the parent species undergoes a large frequency shift (ca. 18 cm⁻¹) only in the IR spectrum of the 1-13C labeled modification, and therefore the v_7 band is assignable to the symmetric CO₂ stretching mode.

The three distinct bands observed at $1471.2 (v_5)$, 1461.4 (v_6), and 1442.4 (v_{18}) cm⁻¹ for the parent molecule may be assigned to two antisymmetric CH₃ deformation modes and one CH₂ bending mode. However, the results of the ¹³C-isotopic shifts suggest that the three normal modes are strongly intermixed or coupled with several group modes. Since the v_5 and v_6 bands are significantly shifted by the ¹³C-substitution of the CO₂ and CH₂ carbon atoms, they may be assigned to coupling modes of the antisymmetric CH₃ deformation (A'), CH₂ bending, and CO₂ symmetric stretching vibrations. The remaining v_{18} band slightly shifts upon 13 Csubstitution of the CO2 and CH2 groups and is tentatively assigned to the antisymmetric CH3 deformation mode (A''). It is noteworthy that the three bands shift only by ca. 1 cm⁻¹ upon ¹³C-substitution of the CH₃ group. Such a feature has been found in the spectra of methylnitrite [8], thioacetamide [9], and pyruvic acid [10], where the bands assigned to

Table 1. Observed frequencies and tentative assignments for the fundamentals of four isotopic species of sodium propionate at 80 K.

Representation C _s	Notation	Approximate description	CH ₃ CH ₂ COONa parent	((1-):COO.	(2-): CH ₂ .	(3-): CH ₃ .) 3- ¹³ C
v_2	sym CH ₂ str.	2952.5	2952.6	2943.5	2950.1	
v_3	sym CH ₃ str.	2937 a	2937.4	2937 a	2932.4	
v_4	asym CO ₂ str.	1563.2	1523.7	1563.7	1563.7	
v_5	asym CH ₃ deform. +	1471.2	1465.6	1468.5	1470.2	
v_6	CH_2 sciss. + sym CO_2 str.	1461.4	1454.0	1453.5	1460.8	
v_7	sym CO ₂ str.	1428.6	1410.4	1428.7	1429.1	
		1417 ^b	1421, 1402 b	1414 ^b	1416 ^b	
v_8	sym CH ₃ deform.	1369.1	1367.7°	1368.4	1360.2	
		1376 ^b		1375 b	1367 ^b	
v_9	CH ₂ wag.	1301.4 °	1298.4°	1286.5	1299.6 °	
v_{10}	$\begin{cases} \text{ in-plane CH}_3 \text{ rock } + \\ \text{CH}_3 - \text{C str.} \end{cases}$	1077.7	1077.7	1061.8	1067.3	
v_{11}	CH ₃ -C str. (asym CCC str.)	1009.9	1000.0	994.5	995.6	
v_{12}	CO ₂ -C str. (sym CCC str.)	881.4	874.9	879.5	879.3	
v_{13}	CO ₂ bend + asym CCC str.	646.7	646.8	637.2	646.8	
v_{14}	$\begin{cases} \text{ in-plane CO}_2 \text{ rock } + \\ \text{sym CCC str.} \end{cases}$	507.4	505.1	505.8	502.7	
v_{15}	CCC bend	291.5	290.8	291.2	289.8	
A"	v_{16}	asym CH3 str.	2970.2	2970.2	2970.0	2960.7
	v_{17}	asym CH ₂ str.	2940.0	2940 a	2933.3	2940.0
	v_{18}	asym CH ₃ deform.	1442.4	1438.6	1438.8	1441.5
	v ₁₉	∫ CH ₂ twist + out-of-plane CH ₃ rock	1250.4	1248.4	1249.8	1245.4
	v_{20}	out-of-plane CH ₃ rock + CH ₂ twist or rock	1080 a	1080 a	1071.4	1075.9
	v_{21}	CH ₂ rock	812.9	805.0	810.5	811.7
	v ₂₂	out-of-plane CO ₂ rock	584.9	576.4	583.9	584.8

a Shoulders.

the methyl antisymmetric deformation modes experience downward shifts of only about 2 cm⁻¹ upon ¹³C-substitution of the methyl groups.

In contrast to the v_5 , v_6 , and v_{18} bands, the v_8 band observed at 1369.1 cm⁻¹ for the parent species undergoes a significant shift to 1360.2 cm⁻¹ upon ¹³C-substitution of the CH₃ group. Thus, by analogy with the CH₃ deformational property of sodium acetate [3] and pyruvate [4], the v_8 band can be reasonably assigned to the symmetric CH₃ deformation mode.

The v_9 band observed at $1301.4 \, \text{cm}^{-1}$ for the parent species, which shifts to $1286.5 \, \text{cm}^{-1}$ upon ^{13}C -substitution of the CH₂ group, may be due to a CH₂ wagging motion. The weak v_{19} band centered

at $1250.4 \, \mathrm{cm^{-1}}$ is tentatively assigned to a $\mathrm{CH_2}$ twisting mode, however a significant methyl $^{13}\mathrm{C}$ -shift (5 cm $^{-1}$) of the ν_{19} band suggests considerable coupling of this vibration to an out-of-plane rocking mode of the $\mathrm{CH_3}$ group (ν_{20} , discussed later).

(3) Fingerprint region, $1200 - 200 \text{ cm}^{-1}$

The main problem of the assignment in this region, in which 6A' + 3A'' fundamentals are to be expected, focuses on two methyl rocking and two skeletal CC stretching modes. An expanded section of the low-temperature spectra for the parent and its three 13 C-isotopic species is shown in Fig. 5 in the frequency region $1100 - 860 \text{ cm}^{-1}$. One strong band

^b Minor components resulting from factor-group splittings.

^c Averaged values of factor-group doublets of similar intensity.

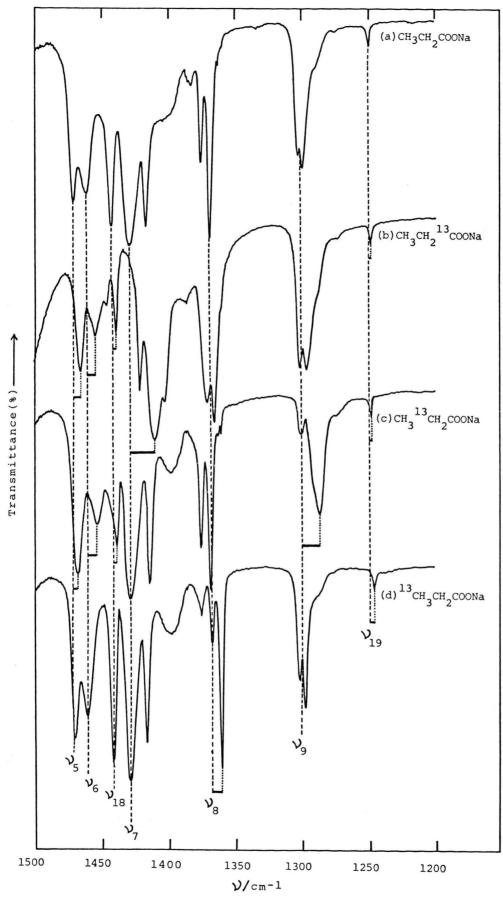


Fig. 4. CO₂-stretching, CH₃- and CH₂-bending region of the spectra of CH₃CH₂COONa (a), CH₃CH₂l³COONa (b), CH₃l³CH₂COONa (c), and 13 CH₂COONa (d) at 80 K; 0.35 mg of sample suspended in 150 mg of KBr. For broken, dotted, and horizontal lines, see Figure 3. For symbols v_i , see the notations given in Table 1.

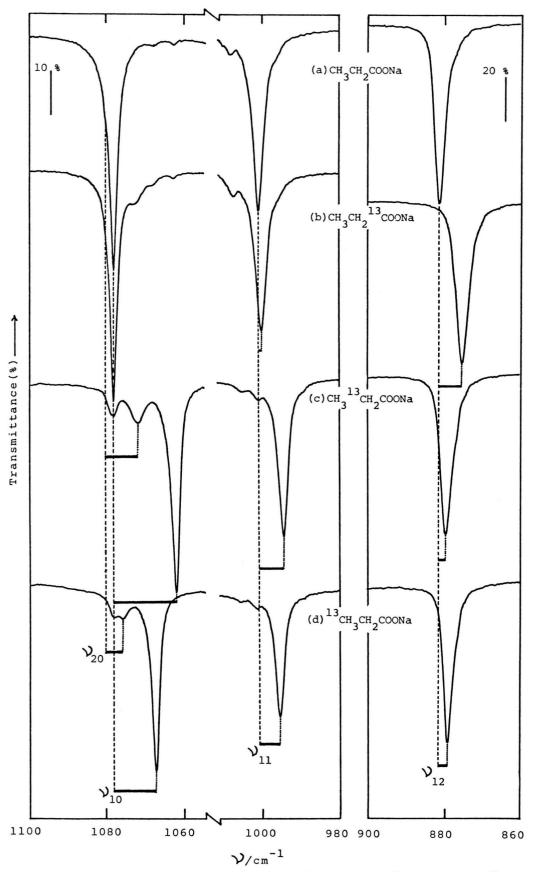


Fig. 5. Fingerprint region of the spectra of CH_3CH_2COONa (a), $CH_3CH_2^{13}COONa$ (b), $CH_3^{13}CH_2COONa$ (c), and $^{13}CH_3CH_2COONa$ (d) at 80 K; 0.35 mg of sample suspended in 150 mg of KBr. For broken, dotted, and horizontal lines, see Figure 3. For symbols v_i , see the notations given in Table 1. Weak bands appearing in a higher frequency side of the spectra (c) and (d) are due to coexisting parent species.

 (v_{10}) with a shoulder (v_{20}) was observed at 1077.7 cm⁻¹ for the parent species, which undergoes large shifts by 15.9 cm⁻¹ and 10.4 cm⁻¹ upon ¹³C-substitution of the CH₂ and CH₃ groups, respectively. Based on this observation of the 13 C-shifts, the v_{10} band can be assigned to a coupled mode having contributions from the in-plane CH₃ rocking and CH₃-C stretching vibrations. The large isotopic shifts of the v_{10} band enable us to locate the v_{20} bands at 1071.4 and 1075.9 cm⁻¹ in the spectra of 2- and 3-13C labeled modifications, respectively (see Fig. 5c and 5d). The v_{20} band could be assigned to a coupled mode of the out-of-plane CH3 rocking and CH2 twist or rocking vibrations.

The v_{11} and v_{12} bands centered at 1000.9 and 881.4 cm⁻¹ for the parent species were tentatively assigned to the CH₃-C and C-CO₂ stretching modes, respectively, by taking into account the characteristic ¹³C-shifts of the fundamentals (see Fig. 5 and Table 1). The assignments of the remaining 5 fundamental bands to the descriptions given in Table 1 were made by comparison with the vibrational assignments of sodium acetate [3] and propane [7], and with the result of a microwave study for propionic acid reported by Stiefvater [11]. It should finally be noted that the vibrational descriptions given for the 5 fundamental bands are only qualitative in the sense that many of the normal modes may be strongly intermixed or coupled with several group modes.

Conclusion

In this paper, we have presented IR fundamental frequeny data for sodium propionate including its three ¹³C-substituted species. The fine structures of complex overlapping regions were revealed by measuring the spectra at liquid N₂ temperature. The low temperature measurements combined with ¹³Cisotopic shifts have been demonstrated to be effective in elucidating vibrational assignments for some fundamentals of sodium propionate. The degree of coupling between several group modes has been clarified by taking into consideration the ¹³C-shifts of the fundamentals.

- [1] B. Dupuy and C. Garrigou-Lagrange, J. Chim. Phys. **62,** 1359 (1965).
- [2] M. Kakihana, M. Kotaka, and M. Okamoto, J. Phys. Chem. **86**, 4385 (1982).
- [3] M. Kakihana, M. Kotaka, and M. Okamoto, J. Phys. Chem. **87**, 2526 (1983).
- [4] M. Kakihana and M. Okamoto, J. Phys. Chem. 88, 1797 (1984).
- [5] International Union of Pure and Applied Chemistry (IUPAC), Tables of Wavenumbers for the Calibration of Infrared Spectrometers"; Pergamon Press, New York 1977.
- [6] L. J. Bellamy, The Infrared spectra of Complex Molecules-Advances in Infrared Group Frequencies, 2nd ed.; Chapman & Hall, New York 1980, Vol. 2.
- [7] R. A. R. Pearce and I. W. Levin, J. Chem. Phys. 70, 370 (1979).
- P. N. Ghosh and H. H. Günthard, Spectrochim. Acta 37 A, 1055 (1981).
- [9] U. Anthoni, P. H. Nielsen, and D. H. Christensen, Spectrochim. Acta 41 A, 1327 (1985).
 [10] H. Hollenstein, F. Akermann, and H. H. Günthard,
- Spectrochim. Acta 34A, 1041 (1978).
- [11] O. L. Stiefvater, J. Chem. Phys. 62, 233 (1975).